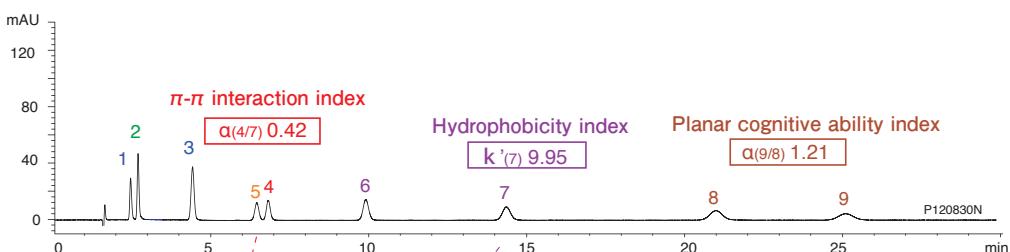
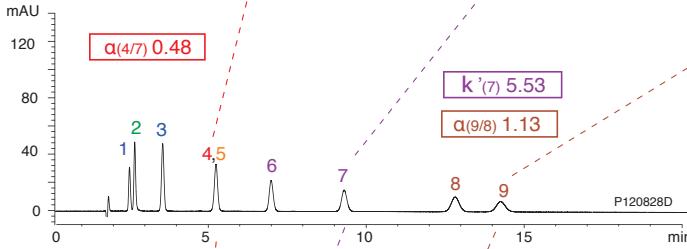


## Comparison of separation selectivity among YMC-Triart series

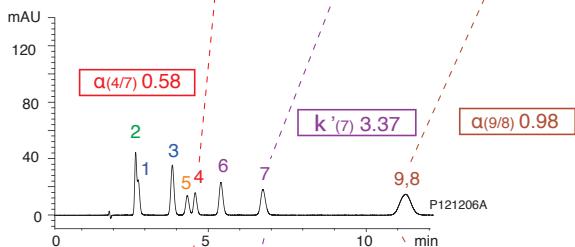


### Triart C8



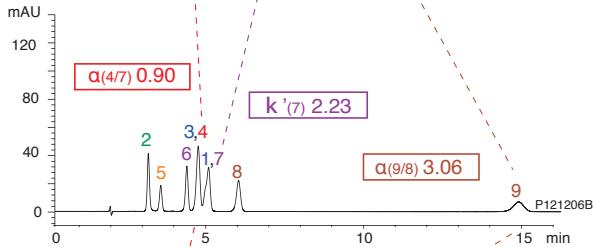
Triart C8 shows similar selectivity to Triart C18, but shorter retention times than C18.

### Triart Phenyl

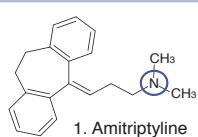


On Triart Phenyl and Triart PFP, π-π interaction and polar interaction as well as hydrophobic interaction contribute to separation. They show different separation selectivity from Triart C18 or Triart C8.

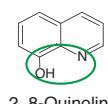
### Triart PFP



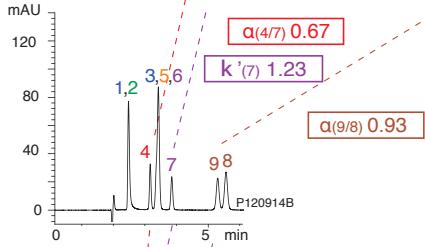
#### Basic compound



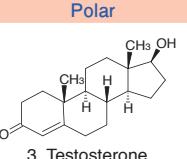
#### Coordination compound



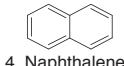
### YMC-Pack Ph



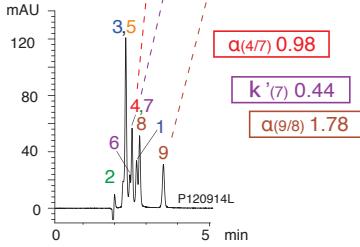
#### Neutral compounds



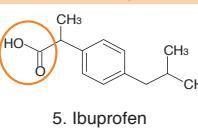
#### π-π interaction



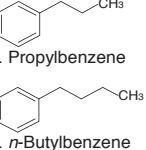
### YMC-Pack CN



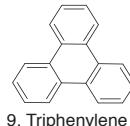
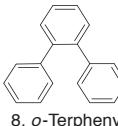
#### Acidic compound



#### Hydrophobicity



#### Planar cognitive ability



Column : 5 μm, 150 X 3.0 mmI.D.  
Eluent : 20 mM H<sub>3</sub>PO<sub>4</sub>-KH<sub>2</sub>PO<sub>4</sub> (pH3.1)/methanol (25/75)  
Flow rate : 0.425 mL/min  
Temperature : 40°C  
Detection : UV at 265 nm  
Injection : 4 μL

A mixture that consists of compounds with various characteristics is analyzed with reversed phase Triart columns. In addition to hydrophobic interaction, secondary interactions such as π-π interaction and polar interaction are different from column to column. Those parameters have great impact on retention capacity ( $k'$ ) and separation factor ( $\alpha$ ). By utilizing the difference in separation characteristics, wide range of compounds can be well-separated with Triart series.